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2-Hydroxy-5-[(*E*)-(4-methoxyphenyl)diazenyl]benzoic acid

T. S. Basu Baul,^a S. Dhar^a and E. R. T. Tiekink^b*

^aChemical Laboratory, Regional Sophisticated Instrumentation Centre, North Eastern Hill University, Bijni Complex, Shillong 793 003, India, and ^bDepartment of Chemistry, The University of Adelaide, Australia 5005 Correspondence e-mail: edward.tiekink@adelaide.edu.au

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The title compound, $C_{14}H_{12}N_2O_4$, shows an *E* conformation about the diazenyl N atoms. The crystal structure features layers of molecules with the primary connection between the layers afforded by carboxylic acid dimer motifs; no evidence for extensive π - π stacking between the layers was found.

Comment

Organotin compounds containing carboxylate ligands similar to that derived from the title compound have received considerable attention recently (Basu Baul & Tiekink, 1998, 1999; Willem et al., 1998). These systems have provided a large number of X-ray quality crystals enabling systematic analyses and hence conclusions to be drawn as to the reasons why different motifs (Tiekink, 1991, 1994) are found for triorganotin carboxylates (Willem et al., 1998). A key feature of some of these structures is the presence of π - π interactions that serve to stabilize the crystal structure. Surprisingly, relatively little is known about the structural chemistry of diazocarboxylic acid derivatives themselves (Moreiras et al., 1980; van der Sluis & Spek, 1990) and in order to ascertain the reasons why $\pi - \pi$ stacking is observed in some organotin derivatives but not in others, the structure of the title compound, (I) (Fig. 1 and Table 1), was investigated.



The conformation about the diazo group is *E* and derived interatomic parameters are as expected. The molecule is basically planar, although there is a slight twist about the diazenyl group, as seen in the N1–N2–C8–C9 and N2–N1–C6–C5 torsion angles of 176.4 (2) and 8.5 (3)°, respectively. The dihedral angle between the two aromatic rings is

5.43 (7)°. The carboxylic acid group is coplanar with the C2–C7 ring [O1-C1-C2-C3-178.2 (2)°], which facilitates the formation of an intramolecular hydrogen-bonding contact, such that $O3\cdots O2$ is 2.630 (2) Å (see Table 2). Intermolecular hydrogen-bonding interactions are also present.

The structure may be described as a hydrogen-bonded array in the *ac* plane that is stacked along **b**; however, the layers are not flat, there being significant corrugation within them. The arrangement of molecules within the layer is best described as being based on the coronene motif (Desiraju & Gavezzotti, 1989). The layers are stabilized in part by weak hydrogenbonding interactions between the hydroxyl-H atom and a symmetry-related methoxy-O atom, such that $O3 \cdots O4^i$ is 3.003 (2) Å [symmetry code: (i) $\frac{1}{2} + x$, y, $\frac{1}{2} - z$]. Contacts of 3.522 (2) Å occurring between translationally related C1 and $C5^{ii}$ atoms throughout the layer are also noted [symmetry code: (ii) x, y, 1 + z]. Connections between the layers are afforded by carboxylic acid dimer formation $[O1 \cdots O2^{iii}]$



The molecular structure and crystallographic numbering scheme for (I). Displacement ellipsoids shown at the 50% probability level (*ORTEPII*; Johnson, 1976).

2.631 (2) Å; symmetry code: (iii) 2 - x, -y, 2 - z]. No evidence was found for extensive $\pi - \pi$ interactions in the lattice. However, between layers, centrosymmetrically related pairs of benzoic acid fragments approach each other at distances indicative of $\pi - \pi$ interactions. The closest contact of 3.295 (2) Å occurs between C1 and C3^{iv}, and the average separation between the least-squares planes through each of the O₂CC₆ fragments is 3.24 Å [symmetry code: (iv) 2 - x, -y, 1 - z]. Presumably, $\pi - \pi$ interactions involving the methoxy-phenyl fragment and hence, the whole molecule, are precluded by the bulky nature of the methyl group. Finally, evidence for a C-H···ring interaction between the layers is also noted, such that C5-H4···Cp^v (where Cp is the centroid of the C2-C7 ring) is 2.78 Å and the angle at H4 is 139° [symmetry code: (v) $x, \frac{1}{2} - y, \frac{1}{2} + z$].

Experimental

Diazo-coupling of *p*-methoxyaniline and *o*-hydroxybenzoic acid in alkaline solution yielded the title compound in 52% yield. Recrystallization from benzene solution gave orange–red crystals (m.p. 482–483 K).

Crystal data

 $\begin{array}{l} C_{14}H_{12}N_2O_4\\ M_r = 272.26\\ \text{Orthorhombic, }Pbca\\ a = 24.902 \ (6) \ \text{\AA}\\ b = 15.109 \ (6) \ \text{\AA}\\ c = 6.674 \ (2) \ \text{\AA}\\ V = 2511 \ (1) \ \text{\AA}^3\\ Z = 8\\ D_x = 1.440 \ \text{Mg m}^{-3} \end{array}$

Data collection

Rigaku AFC-7*R* diffractometer ω scans 6617 measured reflections 2894 independent reflections 1800 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.030$ $\theta_{\text{max}} = 27.52^{\circ}$

Refinement

| Refinement on F | H-atom parameters not refined |
|------------------|--|
| R = 0.034 | $w = 1/[\sigma^2(F_o) + 0.00001 F_o ^2]$ |
| wR = 0.033 | $(\Delta/\sigma)_{\rm max} < 0.001$ |
| S = 1.650 | $\Delta \rho_{\rm max} = 0.24 \text{ e} \text{ Å}^{-3}$ |
| 1800 reflections | $\Delta \rho_{\rm min} = -0.20 \text{ e} \text{ Å}^{-3}$ |
| 181 parameters | |

Mo $K\alpha$ radiation

reflections

 $\mu = 0.107 \text{ mm}^{-1}$

 $\theta = 6.4 - 9.6^{\circ}$

T = 173 K

 $h = 0 \rightarrow 32$

 $k=-1\rightarrow 19$

3 standard reflections

every 400 reflections

intensity decay: -1.39%

 $l = -8 \rightarrow 8$

Cell parameters from 25

Trigonal plate, orange–red $0.45 \times 0.24 \times 0.19$ mm

Table 1

Selected geometric parameters (Å, °).

| O1-C1 | 1.311 (2) | O4-C14 | 1.427 (3) |
|------------|-----------|----------|-----------|
| O2-C1 | 1.241 (2) | N1-N2 | 1.260 (2) |
| O3-C3 | 1.355 (2) | N1-C6 | 1.432 (2) |
| O4-C11 | 1.367 (2) | N2-C8 | 1.424 (2) |
| | | | |
| C11-O4-C14 | 117.1 (2) | O1-C1-O2 | 122.7 (2) |
| N2-N1-C6 | 112.2 (2) | O1-C1-C2 | 115.2 (2) |
| N1-N2-C8 | 115.1 (2) | O2-C1-C2 | 122.1 (2) |
| | | | |

All H atoms were located from a difference map and refined but were fixed in the final cycles of refinement (O-H 0.86 and 0.90 Å; C-H 0.95–1.03 Å).

Data collection: *MSC/AFC Diffractometer Control Software* (Molecular Structure Corporation, 1999); cell refinement: *MSC/AFC Diffractometer Control Software*; data reduction: *TEXSAN* (Molecular Structure Corporation, 1997–1999); program(s) used to solve

Table 2 Hydrogen-bonding geometry (Å, °).

 $D - H \cdot \cdot \cdot A$ D-H $H \cdot \cdot \cdot A$ $D \cdot \cdot \cdot A$ $D - H \cdot \cdot \cdot A$ $O1 - H1 \cdot \cdot \cdot O2^{i}$ 0.90 2.631 (2) 176 1.73 O3−H2···O2 0.86 1.87 2.630(2) 146 O3−H2···O4ⁱⁱ 0.86 2.46 3.003(2)122

Symmetry codes: (i) 2 - x, -y, 2 - z; (ii) $\frac{1}{2} + x$, y, $\frac{1}{2} - z$.

structure: *SIR*97 (Altomare *et al.*, 1997); program(s) used to refine structure: *TEXSAN*; software used to prepare material for publication: *TEXSAN*.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: SK1403). Services for accessing these data are described at the back of the journal.

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